Application of the Rodriguez—Pattenden Photo-Ring Contraction: Total Synthesis and Configurational Reassignment of 11-Gorgiacerol and 11-Epigorgiacerol

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Harald Weinstabl, Tanja Gaich, and Johann Mulzer*

University of Vienna, Währinger Strasse 38, 1090 Vienna, Austria johann.mulzer@univie.ac.at

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A stereospecific photochemical ring contraction was used as the key step in the first total synthesis of the marine pseudopteranyl diterpene 11-gorgiacerol and its 11-epimer. The synthesis allowed the correction of the configurations that had been misassigned in the literature. In addition, some novel pseudopteranyl derivatives have been made.

The furanocembranoids are a vast family of metabolites that have received wide attention in the synthetic, biosynthetic, and pharmacological communities.¹ Among the manifold structural modifications and rearrangements of the furanobutenolide-based cembranoids (I), the photo-induced ring contraction to the pseudopterane skeleton (II) (Scheme 1) is most remarkable.

This rearrangement has been postulated in the biosynthetic formation of **II**, but in vitro, it has been observed only by two groups so far: first by Rodriguez² and later by Pattenden,³ who has also studied the stereochemistry of the reaction in detail. We now report the application of this Rodriguez–Pattenden ring contraction in the first total synthesis⁴ of two metabolites reported as 11-gorgiacerol (= 11-pseudopteranol) (1)⁵ and 11-epigorgiacerol (= 11-epipseudopteranol) (2)⁵⁶ (Figure 1). These pseudopteranoid diterpenes have been isolated from the gorgonian coral *Pseudopterogorgia acerosa* and have been characterized by NMR and mass spectrometry. In particular, the configurations at the carbinol center C-11 were assigned by ¹H-¹H NMR coupling constants only.





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Our synthesis of 1 and 2 has now demonstrated that these assignments were wrong and have to be permuted so that the former 11-gorgiacerol is now 11-epigorgiacerol and vice versa. Thus, Figure 1 shows the former and the corrected configurations.



Figure 1. Original and corrected structures of 11-gorgiacerol and its 11-epimer.

The synthesis (Scheme 2) started with the known⁷ ester acetal **3** (readily available in four steps from (R)-(-)carvone). Reduction to the aldehyde, aldol addition of methyl acetate, and oxidation of the hydroxyl ester afforded keto-ester **4**. Deprotonation and alkylation with iodide **5** furnished keto acetylide **6**, which was cyclized⁸ to the furan under basic conditions. Acid-catalyzed hydrolysis of the acetal led to aldehyde **7** which was subjected to an aldol addition with selenolactone **8**.⁹ Oxidative elimination of the selenium gave butenolides **9a/b**,¹⁰ readily separated by column chromatography. As both epimers were required for the envisaged structural assignment, no efforts were spent to improve the stereoselectivity of the aldol addition.

Ring-closing metathesis $(RCM)^{11}$ with Grubbs' secondgeneration catalyst gave (Z)-olefins **10a** and **10b** stereoselectively (Scheme 2).

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Scheme 2. Total Syntheses of 1 and 2



The synthesis was completed by a photoch emical ring contraction of **10a** and **10b** which furnished the desired pseudopteranes **1** and **2** in acceptable yields. Furan **2** was crystalline, which allowed the unambiguous assignment of its structure by single-crystal X-ray diffraction (Figure 2).

¹H and ¹³C NMR data were in full agreement with the published data (see the Supporting Information).

The optical rotations of our samples differ from the published values but have the same sign (1: $[\alpha]^{20}_{D} = +83.0$ (c = 0.33; CHCl₃) vs +22 (c = 0.1, CHCl₃);⁵ 2: $[\alpha]^{20}_{D} = +130.0$ (c = 0.25; CHCl₃) vs +173.2 (c = 0.12, CHCl₃).⁶

An analogous sequence (Scheme 3) was applied to prepare 11-epigorgiacerol acetate 13 from 9a/b. Crystalline acetate 12b was subjected to a single crystal diffraction (Figure 3). To show that no epimerization occurred at the C-11 center during the photochemical ring contration, compound 2 was subjected to an acetylation reaction. The spectral data of the resulting product were in complete agreement with those of compound 13.

On studying the photoreaction of compound **10b** in more detail, we found, in confirmation of Pattenden's earlier results,³ that the labile (*E*)-isomer **14b** (Scheme 4) was produced first and was isolated after 20% conversion.

⁽⁷⁾ González, M. A.; Ghosh, S.; Rivas, F.; Fischer, D.; Theodorakis, E. A. *Tetrahedron Lett.* **2004**, *45*, 5039–5041.

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Scheme 3. Synthesis of 11-Epigorgiacerol Acetate 13



In the ¹H NMR spectrum of **14b**, considerable line broadening was observed at room temperature because of restricted conformation mobility of the highly strained macrocyclic ring. On heating to 55 °C, the lines sharpened to give a resolved spectrum (see the Supporting Information). On prolonged irradiation, and without reisomerization to 10b, the (E)-olefin 14b underwent ring contraction to 2 with stereoselective generation of the new chiral center at C-7. Obviously, no direct isomerization of 10b to 2 is involved. As pointed out by Pattenden.^{1b,3} the conversion of **14b** into **2** may occur directly via a concerted [1,3]-sigmatropic shift or via a bis-allylic diradical such as 15b in which the original configuration at C-10 (numbering as in Scheme 4) is lost. We do not know the conformation of 14b. However the crystal structures of 2 and 12b (Figures 2 and 3) show that the migrating σ -bond stands virtually perpendicular on the allylic plane. Hence, the conformational situation in such systems is well suited for a suprafacial 1,3-shift, whether concerted or stepwise.12

Aiming for novel pseudopterane derivatives (Scheme 5), acetate 13 was treated with K_2CO_3 to generate methyl ether 17 with complete retention of configuration.

As an S_N1 mechanism is highly unlikely under the conditions, this surprising result is interpreted in terms of an addition/elimination process via 16 (Nu=OMe), in which C-11 is attacked by the strongest nucleophile in the system (=OMe) from the less hindered ring face. By contrast, the reaction of 13 with methanol under acidic conditions furnished adduct 18. Remarkably, only one of the isopropenyl groups was attacked, and the 11-OH was formed by acidic methanolysis of

Scheme 4. Mechanistic Considerations





Figure 2. Crystal Structure of 2.



Figure 3. Crystal structure of 12b.

the acetate. On attempting the dehydration of alcohol **1** with Burgess' reagent,¹³ urethane **21** was obtained

⁽¹²⁾ For similar considerations, see: Kimbrough, T. J.; Roethle, P. A.; Mayer, P.; Trauner, D. Angew. Chem. 2010, 122, 2675–2678. Angew. Chem., Int. Ed. 2010, 49, 2619–2621.

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stereoselectively, which is a close analogue of deoxytobagolide (22).^{14,15} We postulate that 21 is formed from primary adduct 19 via a [1,3]-sigmatropic shift with

(15) The relative configuration of **21** has been safely assigned by appropriate ${}^{1}H-{}^{1}H-NOE$ effects (see the Supporting Information).

cheletropic elimination of SO_3 . The stereochemistry of the rearrangement may be rationalized in terms of a six-membered transition state **20**, in which the nitrogen attacks C-9 from the less hindered side. This assumption is strengthened by the fact that **2** did not undergo rearrangement under the same conditions.

In conclusion, we have achieved (1) a concise synthesis of the pseudopteranyl alcohols 1 and 2 from (*R*)-(–)-carvone in 12 steps (plus two steps for the preparation of selenolactone **8** from (*S*)-tosyl glycidol) in 11% and 8% overall yield; (2) a reassignment of the configuration at C-11 and a confirmation of the absolute configuration of 1 and 2; (3) a confirmation of the mechanism of the Rodriguez-Pattenden ring contraction; and (4) some substrate directed diastereoselective substitutions on the generic core which led to novel pseudopteranyl derivatives.

As for the biological properties of 1 and 2, it has been shown that 2 has cytotoxic activity in the micromolar range, whereas 1 is inactive.⁶ Thus, the behavior of our new pseudopterane derivatives is of interest and will be investigated in ongoing studies.

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Supporting Information Available. Experimental procedures and full characterization of all new compounds including copies of ¹H and ¹³C NMR spectra and crystal structure analysis of **2** and **12b** (CIF). This material is available free of charge via the Internet at http://pubs.acs. org.

The authors declare no competing financial interest.